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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
09/980,321	11/15/2001	Yoshinori Mashiko	4353	9792
75	590 12/22/2004	•	EXAM	INER
Carothers and Carothers			DUONG, THANH P	
445 Fort Pitt Bl				
Pittsburgh, PA	15219		ART UNIT	PAPER NUMBER
-	,		1764	

DATE MAILED: 12/22/2004

Please find below and/or attached an Office communication concerning this application or proceeding.

	Application No.	Applicant(s)
	09/980,321	MASHIKO ET AL.
Office Action Summary	Examiner	Art Unit
	Tom P Duong	1764
The MAILING DATE of this communication		
Period for Reply		·
A SHORTENED STATUTORY PERIOD FOR RE THE MAILING DATE OF THIS COMMUNICATIO - Extensions of time may be available under the provisions of 37 CFF after SIX (6) MONTHS from the mailing date of this communication. - If the period for reply specified above is less than thirty (30) days, a - If NO period for reply is specified above, the maximum statutory per - Failure to reply within the set or extended period for reply will, by state Any reply received by the Office later than three months after the material patent term adjustment. See 37 CFR 1.704(b).	N. R 1.136(a). In no event, however, may a rereply within the statutory minimum of thirt iod will apply and will expire SIX (6) MON atute, cause the application to become AB	eply be timely filed y (30) days will be considered timely. THS from the mailing date of this communication. ANDONED (35 U.S.C. § 133).
Status		
1)⊠ Responsive to communication(s) filed on 08	9 September 2004	
	his action is non-final.	
3) Since this application is in condition for allo		ers, prosecution as to the merits is
closed in accordance with the practice under		
·	,, ., .,	
Disposition of Claims		•
4) Claim(s) 10-21 is/are pending in the applica		
4a) Of the above claim(s) is/are without	drawn from consideration.	
5) Claim(s) is/are allowed.		
6) Claim(s) <u>10-21</u> is/are rejected.		
7) Claim(s) is/are objected to.		
8) Claim(s) are subject to restriction and	d/or election requirement.	
Application Papers		
9)☐ The specification is objected to by the Exam	iner.	•
10) The drawing(s) filed on is/are: a) a		by the Examiner.
Applicant may not request that any objection to t		
Replacement drawing sheet(s) including the con	- · ·	• •
11) The oath or declaration is objected to by the		
Priority under 35 U.S.C. § 119		
<u> </u>	inn miniter 1 - 05 H 0 0 0	
12) Acknowledgment is made of a claim for forea) All b) Some * c) None of:	ign priority under 35 U.S.C. §	119(a)-(d) or (f).
1. Certified copies of the priority docume	ents have been received	
2. Certified copies of the priority docume		nnlication No
_		· · · · · · · · · · · · · · · · · · ·
	•	received in this readonal stage
application from the International Bur	` ','	ropoi vod
* See the attached detailed Office action for a	nst of the certified copies not i	received.
Attachment(s)		
) Notice of References Cited (PTO-892)	4) Interview S	ummary (PTO-413)
Notice of Draftsperson's Patent Drawing Review (PTO-948)	Paper No(s)/Mail Date
N. I Information Displaces Obstantiants (DTO 4440 DTO 600)	ng) 5) Notice of In	formal Patent Application (PTO-152)
 Information Disclosure Statement(s) (PTO-1449 or PTO/SB/ Paper No(s)/Mail Date 	6) Other:	

DETAILED ACTION

Applicants' remarks and amendments filed on September 9, 2004 have been carefully considered. Claim 10 has been amended. Claims 10-21 are pending in this application.

Claim Rejections - 35 USC § 102

The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless -

- (b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States.
- 1. Claims 10-12 are rejected under 35 U.S.C. 102(b) as being anticipated by Japanese Publication 11-106770 (JP '770). Regarding claim 1, JP '770 discloses a method of producing town gas (methane) comprising the steps of : (A) preparing dimethyl ether (DME) as feed stock (Fig. 4); (B) evaporating said dimethyl ether (heated by exhaust gas 1) and; (C) exothermically reforming (5) said dimethyl ether with steam (Abstract) in the presence of catalyst to produce reformed gas consisting mainly methane (Section 0028). Regarding claim 2, JP '770 disclose the quantity of said steam on reforming is within 10/1 to 0.5/1 molar ratio of steam/dimethyl ether (Section 0027). Regarding claim 3, JP '770 discloses the temperature for catalytic reforming of said dimethyl ether (DME) is within 200°C to 600°C (Section 0028).

Claim Rejections - 35 USC § 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

This application currently names joint inventors. In considering patentability of the claims under 35 U.S.C. 103(a), the examiner presumes that the subject matter of the various claims was commonly owned at the time any inventions covered therein were made absent any evidence to the contrary. Applicant is advised of the obligation under 37 CFR 1.56 to point out the inventor and invention dates of each claim that was not commonly owned at the time a later invention was made in order for the examiner to consider the applicability of 35 U.S.C. 103(c) and potential 35 U.S.C. 102(e), (f) or (g) prior art under 35 U.S.C. 103(a).

2. Claim 14 is rejected under 35 U.S.C. 103(a) as being unpatentable over JP '770 in view of Mandelik (3,771,261). JP '770 fails to disclose the DME is divided and supplied to adiabatic fixed bed reactors and the remaining portion of DME to subsequent reactors. Mandelik teaches the majority of the methane gas is fed to a primary reformer to convert methane gas to synthesis gas, and a minor portion is feed to the secondary reformer to convert the remaining methane gas to synthesis gas (Abstract). This method or concept of splitting the feed to

primary and secondary reformers can be applied in the same manner as splitting the DME feed to each reformer as described in the claimed invention to ensure maximum conversion of DME gas to synthesis gas. Thus, it would have been obvious in view of Mandelik '261 to one having ordinary skill in the art to modify the method of producing town gas of JP '770 with splitting DME feed stream to primary and secondary reformers in order to maximize the conversion of DME to synthesis gas.

3. Claims 13 and 16 are rejected under 35 U.S.C. 103(a) as being unpatentable over JP '770 in view of Japanese Publication 56-002919 (JP '919). Regarding claim 13, JP '770 fails to disclose dimethyl ether (DME) is supplied serially to adiabatic fixed bed reactors through cooling means installed between said reactors. JP '919 discloses coolers (7,9) between reformers (5) and (13) as shown in Fig. 1 to cool the reforming gas prior art to feeding to the methanator. Thus, it would have been obvious in view of JP '919 to modify the method of producing town gas of JP '770 with cooler means as taught by JP '919 in order to cool the reforming gas prior to feeding to the methanator. Regarding claim 16, JP '770 fails to disclose carbon dioxide by-produced by said reforming of said DME is removed from said reforming gas after reforming said DME. JP '919 teaches the reforming gas is feed to the carbon dioxide absorber (18) to remove carbon dioxide in order to obtain a rich methane gas product. Thus, it would have been obvious in view of JP '919 to one having ordinary skill in the art to modify the method of producing town gas of JP '770 with feeding the reforming

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gas to carbon dioxide absorber to remove carbon dioxide in order to obtain a rich methane gas product.

- 4. Claims 16-17 and 20 are rejected under 35 U.S.C. 103(a) as being unpatentable over JP '770 in view of Faust et al. (4,171,343). Regarding claims 16 and 17, JP '770 fails to disclose carbon dioxide by-produced by said reforming of said DME is removed from said reforming gas after reforming said DME, and the absorption solution of aqueous alkanolamine and potassium carbonate. Faust teaches the reforming gas is fed to a CO2 removal system 16, which utilizes aqueous triethanolamine and potassium carbonate solution (Col. 4, lines 21-30) to remove CO2 prior to feeding to the methanator (17). Thus, it would have been obvious in view of Faust '343 to one having ordinary skill in the art to modify the method of producing town gas of JP '770 with CO2 removal system with aqueous alkanolamine and potassium carbonate as taught by Faust in order to remove CO2 by-product. Regarding claim 20, Faust discloses the reforming gas comprising of hydrogen, carbon monoxide, and carbon dioxide (Col. 3, line 55) and these components are converted into methane by a methanator (17).
- 5. Claims 14, 16, 18, and 20 are rejected under 35 U.S.C. 103(a) as being unpatentable over JP '770 in view of Winter (4,613,492). Regarding claim 14, JP '770 fails to disclose the DME feed is divided and supplied to adiabatic fixed bed reactors and the remaining portion of DME to subsequent reactors. Winter '492 teaches the important to bypassing a portion of the feed around the primary

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reformer to the secondary reformer(col. 1, lines 65-70) to reduce the overall cost of the reforming system and operating cost (Col. 2, lines 3-6). The method or concept of bypassing a portion of the hydrocarbon gas around the primary reformer to the secondary reformer can be applied in the same manner as splitting the DME feed to each reformer as described in the claimed invention so that the equipment and operating costs of the reforming system is minimized. Thus, it would have been obvious in view of Winter to one having ordinary skill in the art to modify the method of producing town gas of '770 with bypassing a portion of the hydrocarbon gas or DME gas around the primary reformer to secondary reformer as taught by Winter in order to reduce the equipment and operating costs. Regarding claims 16 and 18, JP '770 fails to disclose carbon dioxide by-produced by said reforming of said DME is removed from said reforming gas after reforming said DME, and the carbon dioxide is removed from said reforming gas by adsorption by a pressure swing method. Winter '492 teaches carbon dioxide by-product is removed from the reformed gas by a pressure swing adsorption (PSA). Thus, it would have been obvious to one having ordinary skill in the art to modify the method of producing town gas of JP '770 with PSA as taught by Winter '492 in order to removal undesirable CO2 byproduct. Regarding claim 20, Winter discloses the reforming gas comprising of hydrogen, carbon monoxide, and carbon dioxide (Col. 1, lines 20-23) and these components are converted into methane by a methanator (Fig. 1).

- 6. Claims 16 and 19-20 are rejected under 35 U.S.C. 103(a) as being unpatentable over JP '770 in view of Ho (6,579,331). Regarding claims 16 and 19, JP '770 fails to disclose carbon dioxide by-produced by said reforming of said DME is removed from said reforming gas after reforming said DME, and the carbon dioxide is removed from said reforming gas by selective separation membrane. Ho '331 teaches the use of a CO2 selective membrane as shown in Figure 2 to purify the reforming gas (Col. 4, lines 18-23) by removing CO2 content prior feeding to the methanator to generate town gas (methane). Thus, it would have been obvious in view of Ho '331 to one having ordinary skill in the art to modify of producing town gas of JP '770 with CO2 selective membrane to remove undesirable CO2 by-product from the reforming gas. Regarding claim 20, Ho discloses shows on Figure 2 the reforming gas comprising of hydrogen, carbon monoxide, and carbon dioxide and these components are converted into methane by methanator (Fig. 2).
- 7. Claim 21 is rejected under 35 U.S.C. 103(a) as being unpatentable over JP '770 in view of Japanese Publication 11-293263 (JP '263). JP '770 fails to disclose a portion of said DME is added to said reformed gas as carburant. JP '263 teaches DME is mixed with natural gas (methane gas) as a carburant agent (Section 0013) in order to increase the calorific value or heat value of the town gas (Section 0017). Thus, it would have been obvious in view of JP '263 to one having ordinary skill in the art to modify the production of town gas of JP '770

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with mixing DME with methane gas in order to increase the heat value of the town gas.

Response to Arguments

Applicant's arguments filed 9/16/04 have been fully considered but they are not persuasive. (1) Applicants' argument that JP '770 does not disclose an exothermically DME reforming process. Examiner respectfully disagrees since the DME reforming process of JP '770 is the same as the claimed invention. It is known in the art that typically, the reforming process initially required steam or heat to initiate the reaction; however, the reforming reaction itself is an exothermic reaction and JP '770 discloses a reforming process with exothermic reactions. (2) With respect to the byproducts of the reforming process, JP '700 clearly discloses methane as one of the reforming products with reaction temp. of 200-500°C. (3) With respect to the argument of steam/DME ratio, JP '770 discloses the steam/DME molar ratio of 1-10 (Section 0027). (4) With respect to suppressing the temperature rise in the reaction system, JP '770 discloses the reforming temperature is 200 to 250°C and a higher temperature greater than 500°C will results in a higher undesirable byproducts. Thus, it is clear that JP '770 desires to suppress the reaction temperature in the same manner as the claimed invention.

Conclusion

THIS ACTION IS MADE FINAL. Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the mailing date of this final action.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Tom P Duong whose telephone number is (571) 272-2794. The examiner can normally be reached on 8:00AM - 4:30PM.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Glenn Caldarola can be reached on (571) 272-1444. The fax phone number for the organization where this application or proceeding is assigned is 703-872-9306.

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Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).

Tom Duong December 13, 2004

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Glenn Caldarola Supervisory Patent Examiner Technology Center 1700